

# Na<sub>2</sub>IrO<sub>3</sub> as a molecular orbital crystal

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Spin-orbit (SO) coupling can lead to many nontrivial effects in such fields as spintronics (Rashba effect), topological insulators, or to the formation of topologically protected states in systems described by the Heisenberg-Kitaev model, recently proposed for Na<sub>2</sub>IrO<sub>3</sub>. This proposal is based on the fact the SO coupling for Ir is very strong. We show, however, that Na<sub>2</sub>IrO<sub>3</sub> represents a highly unusual case, in which the electronic structure is dominated by the formation of quasi-molecular composite orbitals (QMOs). The QMOs consist of six atomic orbitals on an Ir hexagon, and the orbital moment of each QMO is quenched. The concept of such QMOs in solids is completely new, and invokes very different physics compared to the models considered previously. For instance, one has to account for Hubbard correlations among the QMOs, and not individual atomic orbitals. Both, the insulating behavior and the experimentally observed zigzag antiferromagnetism in Na<sub>2</sub>IrO<sub>3</sub> naturally follow from the QMO model.

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The recent discovery of quasihexagonal sodium iridate Na<sub>2</sub>IrO<sub>3</sub>, see *e.g.* Ref. 1 with a honeycomb lattice of Ir<sup>4+</sup> (5d<sup>5</sup>) ions has generated enormous interest. A very nontrivial hierarchy of energy scales, with spin-orbit coupling exceeding the three energies typically dominating the physics of 3d and 4d transition metals (the Hubbard  $U$ , the Hund's  $J_H$ , and the one-electron hopping  $t$ ), has been proposed theoretically [2], and shown to have a highly nontrivial phase diagram [2, 3]. It is presumed in this so-called Heisenberg-Kitaev model that the three orbitals forming the  $t_{2g}$  band of Ir  $d$  states are nearly degenerate in energy and their hopping between sites is limited by the nearest neighbor (n.n.) bonds and proceeds through two main channels. One is the direct overlap of the Ir  $d$  orbitals,  $t_1$ , and the other an assisted hopping through oxygen  $p$  orbitals,  $t'_1 = t_{pd\pi}^2/(E_F - E_p)$ , where  $E_F$  is the Fermi energy lying in the  $d$  band, *i.e.*  $E_F \sim E_d$ . For ideal IrO<sub>6</sub> octahedra this latter hopping is allowed, for each of the three n.n. bonds, only between a particular pair of unlike orbitals, for instance, between  $d_{xz}$  and  $d_{yz}$ . It is further assumed that the spin-orbit interaction is much larger than any one-electron hopping. If that were the case, the degeneracy of the  $t_{2g}$  orbitals would have been lifted and the ground state would be a Kramers doublet with total angular momentum (effective spin) 1/2. From the one-electron point of view, this means that the three real harmonics do not represent a good basis to describe the  $t_{2g}$  states; a better basis would be one that maximizes the angular momentum [4, 5]:  $d_{xz} + id_{yz}$ ,  $d_{yz} + id_{xy}$ , and  $d_{xy} + id_{xz}$ . A strong spin-orbit will hold the spins of these complex orbitals parallel to  $z$ ,  $y$ , and  $x$ , respectively. If only hopping *via* oxygen  $p$  orbitals is allowed (see a detailed discussion later in this paper), and the Ir-O-Ir angles are all 90°, then for each of these complex orbitals there will be one particular bond direction where the effective hopping will be maximized.

According to the Goodenough-Kanamori rules, in this case the magnetic coupling along this bond will be ferromagnetic, and maximized when the spins are parallel to the corresponding (different for each bond) quantization axis, with the coupling strength of the order of  $J_O t_{pd\pi}^4/(E_F - E_p)^4 = J_O [t'_1/(E_F - E_p)]^2$ , where  $J_O$  is the Hund's rule coupling on oxygen. Combined with the direct Heisenberg exchange of the order of  $t_1^2/U$ , these considerations lead to the following Hamiltonian, usually called “Heisenberg-Kitaev” [2]:

$$H_{ij} = J_A \mathbf{S}_i \cdot \mathbf{S}_j - J_F S_i^\gamma S_j^\gamma, \quad (1)$$

where  $S_i$  is the effective spin formed at site  $i$ . For each pair  $i, j$  there exists one particular cubic direction  $\gamma = x, y$  or  $z$ . Here  $J_A > 0$  is the Heisenberg antiferromagnetic term and  $J_F > 0$  is the ferromagnetic superexchange [8]. This Hamiltonian leads to a nontrivial phase diagram, including a quantum spin liquid phase with gapless excitations of Majorana fermion type at large  $J_F$  [6]. Recently, a modification of this model with considerable 2nd and 3rd n.n. Heisenberg exchange interactions has also been proposed [7] mainly because the original model failed to describe the existing experimental data.

This model (Eq. (1)), however, implies a very particular hierarchy of energies, namely the spin-orbit interaction must dominate over both the one-electron hopping and the trigonal crystal field splitting of the  $t_{2g}$  orbitals. Also, in order to ensure a long range magnetic order in this model the superexchange interaction should be comparable with the direct exchange.

It appears that both these conditions are violated in Na<sub>2</sub>IrO<sub>3</sub>. It has been pointed out [9] that the real, accurately determined crystal structure of Na<sub>2</sub>IrO<sub>3</sub> [9, 10] features Ir-O-Ir bond angles that are close to 98°. Note that for 5d metals  $U \sim 1 - 1.5\text{eV} \ll (E_F - E_p)$ . In this regime the ferromagnetic part of the superexchange is

determined by the ratio  $J_O[t'_1/(E_F - E_p)]^2$ , while the antiferromagnetic part by the ratio  $(t'_1)^2/U$ . Given that  $J_O \sim U \ll (E_F - E_p)$ , it becomes obvious that the net interaction is only ferromagnetic in a narrow range of angles close to  $90^\circ$ , where the antiferromagnetic superexchange disappears [11]. An angle of  $98^\circ$  is clearly outside this range.

Secondly, the total width,  $W$ , of the  $t_{2g}$  band is about 1.5 eV, which is larger than the spin-orbit coupling  $\lambda \lesssim 0.5$  eV thus violating the first condition of the Heisenberg-Kitaev model as well.

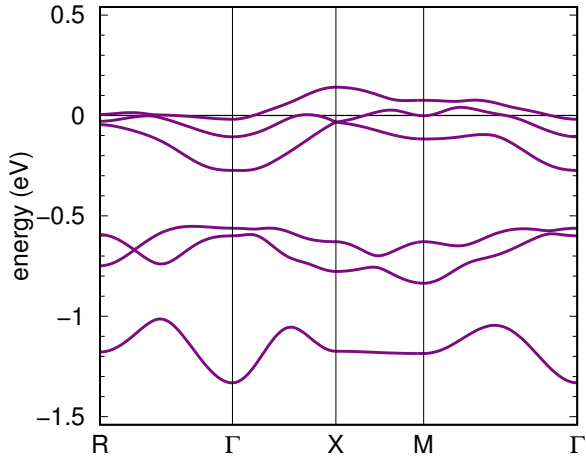


FIG. 1: (a) Electronic structure for the non-magnetic state of  $\text{Na}_2\text{IrO}_3$  without inclusion of spin-orbit coupling for the experimentally determined crystal structure. The calculations were performed with the full potential local orbital (FPLO) basis using the generalized gradient approximation (see Suppl. Information).

Note that the spin-orbit coupling  $\lambda$  is largely reduced by hybridization with oxygen and, as explained below, by the quenching of the orbital moment. The trigonal splitting  $\Delta_T$  could be in principle large, but appears small because of cancellation of various factors, as shown below.

How then should one approach the electronic structure of this compound, given that the energy hierarchy is in reality  $W > U > \lambda > J_H > \Delta_T$ ?

In order to answer this question, we have first performed *ab initio* calculations [12] of the electronic structure, using the experimentally determined  $C2/m$  structure [9], without including the spin-orbit coupling (Fig. 1). A cursory glance at the calculated band structure reveals a picture rather removed from the naive quasiatomic considerations underlying Eq. (1): instead of the expected splitting of the 6  $t_{2g}$  states (there are two Ir atoms per unit cell) into two  $a_{1g}$  and four  $e'_g$  states, we observe an ordering of the states into a single-state band at  $\sim -1.2$  eV, a two-states one at  $\sim -0.6$  eV, and a three-states manifold at the Fermi level, which, as we

TABLE I: Six quasi-molecular orbitals formed by the six  $t_{2g}$  atomic orbitals on a hexagon. Here  $\omega = \exp(i\pi/3) = \frac{1}{2} + i\frac{\sqrt{3}}{2}$ ,  $\omega^2 = \exp(2i\pi/3) = -\frac{1}{2} + i\frac{\sqrt{3}}{2} = -\omega^*$ ,  $\omega^3 = -1$ ,  $\omega^4 = \exp(4i\pi/3) = -\frac{1}{2} - i\frac{\sqrt{3}}{2} = -\omega$ ,  $\omega^5 = \exp(5i\pi/3) = \frac{1}{2} - i\frac{\sqrt{3}}{2} = \omega^*$ ,  $\omega^6 = 1$

Symmetry	Eigenenergy	Eigenvector(s)
$A_{1g}$	$-2t'_1$	1,1,1,1,1,1
$E_{1u}$ (twofold)	$-t'_1$	1, $\omega$ , $\omega^2$ , $-1$ , $\omega^4$ , $\omega^5$ 1, $\omega^5$ , $\omega^4$ , $-1$ , $\omega^2$ , $\omega$
$E_{2g}$ (twofold)	$t'_1$	1, $\omega^2$ , $\omega^4$ , 1, $\omega^2$ , $\omega^4$ 1, $\omega^4$ , $\omega^2$ , 1, $\omega^4$ , $\omega^2$
$B_{1u}$	$2t'_1$	1,-1,1,-1,1,-1

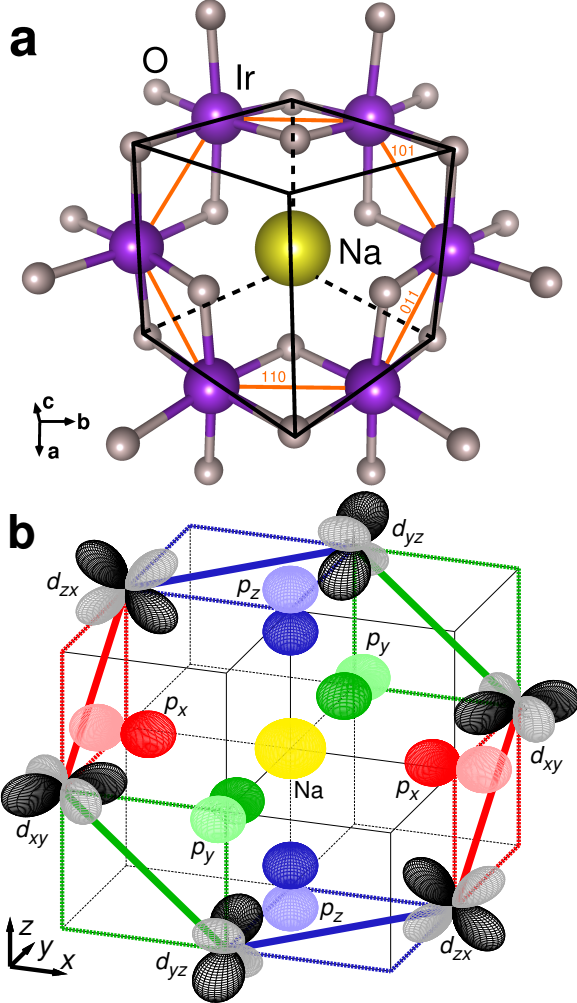
shall see later, consists of slightly overlapping two-states and single-state bands.

To analyze this band structure, we have inverted it using the Wannier function formalism (see Suppl. Information), and extracted the hopping parameters up to the 3rd neighbors. We have found that by far the largest single parameter is the O-assisted n.n. hopping,  $t'_1 = 262 \pm 2$  meV. Therefore, a sensible starting point is the electronic structure obtained by retaining only  $t'_1$  and setting all other parameters to zero.

Let us first look at the crystal structure of  $\text{Na}_2\text{IrO}_3$  in a cubic setting, as shown in Fig. 2(a). The three types of Ir-Ir bonds are parallel to the cubic directions (110), (101), and (011). We call them, respectively,  $xy$ ,  $xz$  and  $yz$  bonds. As illustrated in Fig. 2(b), the only allowed Ir-O  $pd\pi$  hopping path along, for instance, the  $xy$  bond (blue lines) takes us from the  $d_{xz}$  orbital of one Ir to the  $d_{yz}$  one of a neighboring Ir, and so on. Let us now consider an individual orbital, say,  $d_{xy}$ , on a site that in Fig. 3 is labeled as “1”. An electron can only hop along the  $yz$  (red line) bond, into the  $d_{xz}$  state on site 2, or along the  $xz$  (green line) bond, into the  $d_{yz}$  state on site 6. Following the allowed hoppings we realize that an electron starting in a particular orbital state will be always localized on one hexagon, as there are no hoppings in the system that would mix the states on two hexagons. The quasi-molecular composite orbital so formed can be written down as a linear combination of the 6 states on a particular hexagon; the problem is mathematically similar to the molecular  $\pi$ -orbitals of benzene, or to the hypothetical  $\text{SH}_6$  molecule considered in the text book [13], except that here the quasi-molecular orbital is formed by three alternating atomic orbitals. As discussed in Ref. [13], the exact solution for the possible phase combinations of the component orbitals yields four eigenstates as shown in Table 1.

Thus, we have a new basis for constructing our wave functions: instead of two sites per cell, with the total of 6 atomic  $t_{2g}$  orbitals, we have one hexagon per cell (which

FIG. 2: (a) Crystal structure of  $\text{Na}_2\text{IrO}_3$  in the cubic setting. The hexagonal direction is along the 111 direction in this setting. Ir, O and Na atoms are shown as grey, magenta, and yellow spheres, respectively. The three inequivalent Ir-Ir bonds are labeled according to their cubic directions. (b) A cartoon illustrating formation of a quasi-molecular orbital on an  $\text{Ir}_6$  hexagon. If only hopping *via* O  $p$  states is allowed, for each of the three Ir-Ir bond types only hopping between two particular  $t_{2g}$  orbitals is possible. The same holds for the second and third nearest neighbor hopping *via* O  $p$  and Na  $s$  orbitals. Ir-Ir bonds are color coded as follows:  $xy$  bonds are shown by blue lines,  $xz$  bonds by green, and  $yz$  bonds by red ones.



can be identified by the Na atom in the center), with 6 composite orbitals per "site" (hexagon).

The next most important hoppings are second n.n. ones between unlike orbitals, which proceed *via* the two closest oxygen  $p$  orbitals and the diffuse Na  $s$  orbital. These become particularly important if the O octahedra around Ir are trigonally squeezed (as they are). According to our calculation, these hoppings are 79-80 meV,

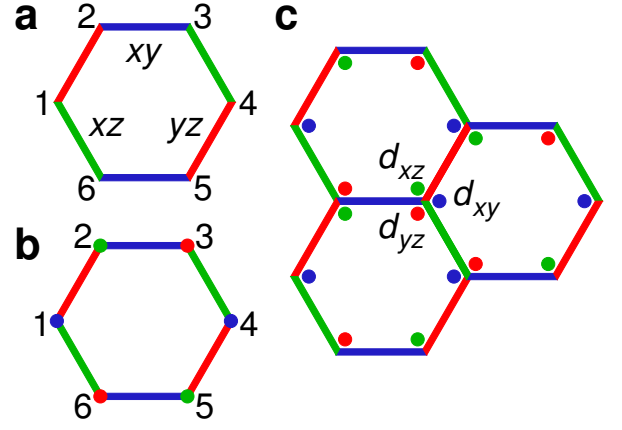


FIG. 3: (a) Schematic plot of a  $\text{Ir}_6\text{Na}$  hexagon. We use the same color coding as in Fig. 2,  $xy$  bonds are shown by blue lines and  $d_{xy}$  orbitals by blue dots, *etc.* (b) A quasi-molecular composite orbital on a given hexagon. (c) Three neighboring quasi-molecular orbitals.

3.3 times smaller than  $t'_1$ . To the lowest order their amplitude is given by  $t'_2 \sim t_{pd\pi}^2 t_{sp}^2 / (E_F - E_p)^2 (E_s - E_F)$ , where the index  $s$  refers to Na  $s$  orbital. Importantly, these hoppings take us from the  $d_{xy}$  orbital on site 1 to the  $d_{yz}$  orbital on site 3, *etc.*, again leaving the electron within the same composite orbital. All other hoppings, including the direct  $dd\sigma$  overlap implied in Eq. (1), are at least a factor of two smaller. [14]

It is interesting that the trigonal splitting  $\Delta$ , which also couples the n.n. QMOs, is small ( $< 25$  meV) even though the distortion itself is large. This can be understood from the fact that three factors of comparable magnitude determine the value of  $\Delta$  [15]: the ratio between the iridium-oxygen  $t_{pp\sigma}/t_{pp\pi}$  (the sign of the ligand field depends on this ratio), the electrostatic field on the Ir site, and the fact that the  $a_{1g}$  orbital extends over towards the positively charged Na (see Suppl. Information), and is thus pushed down compared to the  $e'_2$  orbitals. Apparently in the actual crystal structure of  $\text{Na}_2\text{IrO}_3$  these three factors largely cancel out.

Looking at the actual *non-relativistic* bands of  $\text{Na}_2\text{IrO}_3$ , we see that the overall structure deduced in the previous paragraphs survives (Fig. 1). The lowest band is singly degenerate and separated from the next one. The next lowest band is doubly degenerate, as expected. However, the two upper bands ( $E_{2g}$  and  $B_{1u}$ ) slightly overlap. At the energy where they overlap a peak in the DOS appears (Fig. 4) because of flat band dispersion in that energy range. This peak has interesting consequences. In the same figure we show the calculated non-relativistic DOS in the ferromagnetic state. Since the non-magnetic Fermi level appears close to this peak, the system is unstable against ferromagnetism and exchange splitting of the peak. The high DOS drives the system into the half-metallic state where the majority bands are

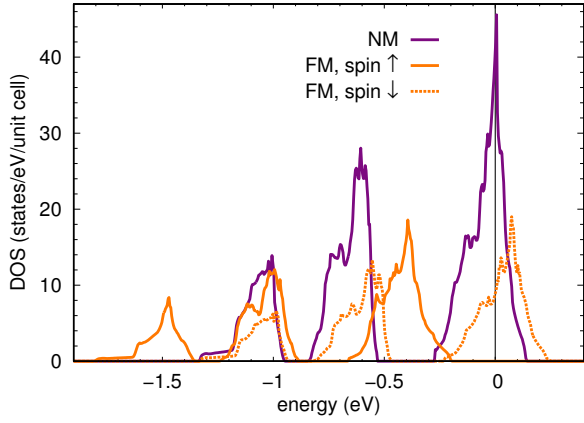


FIG. 4: Non-magnetic (purple) and ferromagnetic (orange) density of states (DOS) of  $\text{Na}_2\text{IrO}_3$  calculated with the FPLO basis using the generalized gradient approximation without spin-orbit coupling (see Suppl. Information).

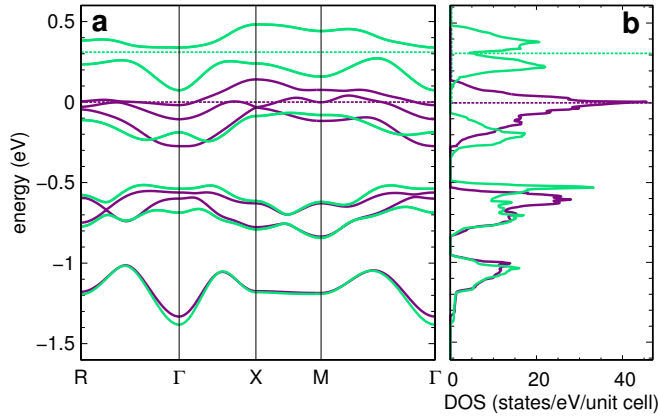


FIG. 5: Effect of spin-orbit on the band structure (a) and DOS (b) of  $\text{Na}_2\text{IrO}_3$ . The purple and green lines refer to the non-magnetic calculation without and with spin-orbit interaction, respectively. Note that the Fermi levels (shown by the horizontal dashed lines) are not aligned.

fully occupied and the Fermi level cuts the minority band right in the middle of the  $E_{2g}$  band. Obviously, this is a rather fragile situation, related to a particularly high DOS at the Fermi level in the non-relativistic calculations, and one may expect the spin-orbit interaction to change the magnetic ground state significantly just by modifying the DOS.

Now that we have established the mechanism of formation of the *non-relativistic* band structure, let us see how turning on the spin-orbit coupling affects it. Here another surprise awaits us. As Fig. 5 shows, the effect is rather moderate. It is obvious that the spin-orbit interaction is strong but not dominant. Moreover, the lower two manifolds, well isolated, experience only a very minor spin-orbital effect, not more than 50 meV, and only a few

meV on average. The upper subbands, as mentioned previously, overlap and mix, which leads to sizable changes, from 0.2 to 0.5 eV. Interestingly, the overlapped manifold of two  $E_{2g}$  and one  $B_{1u}$  bands has now essentially split into three individual subbands (which one cannot any longer, strictly speaking, call  $E_{2g}$  and  $B_{1u}$ , but the notion of the QMOs involving 6 atomic orbitals each still holds). Now it becomes virtually impossible to produce a half-metallic  $S = 1/2$  state, as in the non-relativistic case: an exchange splitting of  $\sim 0.5$  eV, corresponding to  $M = 1 \mu_B$ , is smaller than the combined band width of the upper two bands.

All these seemingly unexpected results are in fact natural consequences of the QMO picture. Indeed, let us consider first the  $A_{1g}$  band (see Table I), and limit ourselves to the  $t'_1$  only model (dispersionless bands). This QMO has the same phases on all 6 sites. Let us consider the spin-orbit interaction  $\hat{\mathbf{L}} \cdot \hat{\mathbf{S}}$  at a given  $\mathbf{k}$ -point, between two QMOs. Let us say, to be specific, that they share an “ $xy$ ” bond, between the sites “2” and “3” in Fig. 3. Let us further say that the distance between the centers of the two QMOs is  $\mathbf{R}$ , so that the phase factor between the two QMOs is  $\exp(i\mathbf{k}\mathbf{R})$ . The two orbitals on sites 2 and 3 are  $d_{xz}$  and  $d_{yz}$  (the first QMO) and  $d_{yz}$  and  $d_{xz}$  (the second QMO), respectively. The matrix elements of the orbital moment operator  $\hat{\mathbf{L}}$  on site 2 and 3 is then

$$\mathbf{L}_2 = \mathbf{L}_3^* = \left\langle d_{xz} + \exp(i\mathbf{k}\mathbf{R})d_{yz} \right| \hat{\mathbf{L}} \left| d_{xz} + \exp(-i\mathbf{k}\mathbf{R})d_{yz} \right\rangle. \quad (2)$$

Since there is another QMO, located at  $-\mathbf{R}$ , and involving the same atomic orbitals on sites 5 and 6, the corresponding matrix elements also need to be added.

A direct calculation shows that  $\mathbf{L}_2 \neq 0$ ,  $\mathbf{L}_2 = \mathbf{L}_3 = -\mathbf{L}_5 = -\mathbf{L}_6$ , and their contributions cancel identically. A bit more tedious arithmetic shows that the two-fold  $E_{1u}$  state also has zero matrix elements of the angular momentum, both diagonal and off-diagonal.

This is not the case for the matrix elements between the  $B_{1u}$  and any of the  $E_{2g}$  states, forming the three upper bands. Acting the same way, one can show that the sum of all four components is nonzero, and, indeed, in the calculations these states are strongly affected by the spin-orbit coupling. As a consequence, the manifold of the three bands splits into three individual bands; their dispersion also increases. This broadening of the upper bands is the factor preventing the system from getting fully spin-polarized in the ferromagnetic calculations, and reducing substantially the energy gain in the magnetic state.

Importantly, as a result of the hybridization between the  $E_{2g}$  and the  $B_{1u}$  bands, the latter becomes fully separated with only a tiny indirect overlap. [16] Since a gap or pseudogap at the Fermi level is advantageous from the point of view of the one-electron energy, one can expect that such a non-magnetic band structure should promote magnetic patterns that do not destroy the pseudogap. As

we shall see, the stripy pattern (and, of course, the ferromagnetic one) destroy the pseudogap, but the zigzag pattern preserves it (see Fig. 6). This gives the latter a serious energetic advantage.

Let us now discuss the possibility of a Mott-Hubbard state in this system. In previous works sometimes values of  $U$  of 3 eV and larger were considered [17, 18]. This is an unjustifiably large number, appropriate for 4d, but not 5d metals. Generally speaking,  $U$  is reduced by about 1 eV when going to 4d metals (a typical value being 2-3 eV) and by about as much when moving to the fifth period. A quasiatomic estimate of  $U$  and  $J$  in  $\text{Na}_2\text{IrO}_3$  using the recipe from Ref. [19], yields  $U = 1.68$  eV and  $J = 0.29$  eV. With these values,  $U - J$  is not enough to drive the whole  $t_{1g}$  band into a Mott regime, because its width  $W$  is larger than  $U$ . On the other hand, the individual subbands are much narrower, and can be strongly correlated in the Mott-Hubbard sense. In this case, however, the relevant energy is not the atomic  $U$ , but  $U_{QMO}$ , the energy cost of placing an extra electron on a QMO. This is similar to correlation effects in the narrow  $t_{1u}$  band in fullerenes [20]. Of course,  $U_{QMO}$  is much smaller than  $U$ ; assuming no additional screening by other QMOs,  $U_{QMO} \sim U/6 \lesssim 0.3$  eV. While this seems like a small number, it is to be compared to the band width of the individual subbands, which for the top three bands is also 0.2-0.3 eV. This situation is very similar to the Mott state in  $\text{TaS}_2$ , where the CDW combined with the spin-orbit interaction creates an isolated very narrow band, which experiences a Mott transition even though  $U$  is only a small fraction of an eV.

The  $U_{QMO}$ , in the non-magnetic state, can enhance the already existing, but extremely small gap in the fully relativistic DFT calculations (the minimal indirect gap is zero within the computational accuracy, while the average direct, *i.e.*, optical gap is about 150 meV), however it would be probably misleading to call the material a Mott insulator, since it is already nonmetallic in DFT when including spin-orbit coupling. This fact has been recently also pointed out in Ref. 18.

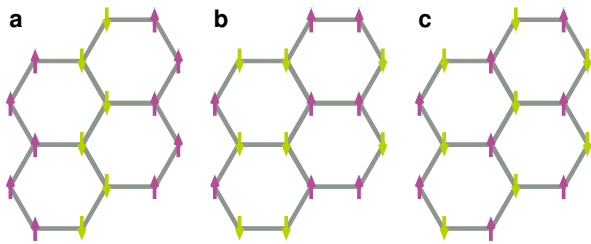


FIG. 6: Three antiferromagnetic patterns considered in this paper: (a) zigzag, (b) stripy, and (c) Néel.

It has been common to try to describe the magnetic ground state in terms of n.n. (up to 3rd neighbors) exchange of various kinds [7]. Given enough parameters,

any given state can be reproduced. In this capacity, mapping effective interactions onto an effective Heisenberg model has been successfully applied to materials where electrons are not localized at all and the concept of superexchange is not applicable (metal Fe, or Fe-based superconductors, for instance). We shall adopt however a different approach and try to speculate about possible magnetic orderings in terms of QMOs, which, as we have established, play a crucial role in the electronic structure of  $\text{Na}_2\text{IrO}_3$ . Since the non-magnetic ground state is insulating, and the magnetic moment is small, one naturally expects an antiferromagnetic ordering. To find out which orders are naturally compatible with the QMO picture, let us consider possible antiferromagnetic patterns and simply calculate the energy of the highest (unoccupied) QMO for a given pattern. On the one-electron energy level, the pattern where the highest level has the largest energy will win (because its occupied QMOs, combined, will have the lowest energy).

The Hamiltonian we need to diagonalize is

$$\Delta_{ex}s_i\delta_{ij} + t\delta_{i+1,j} \quad (3)$$

where  $i = \{1...6\}$ , and  $s_i$  takes +1 or -1, depending on the pattern (Fig. 6).  $\Delta_{ex}$  is the exchange splitting. Specifically, in the Néel state  $s_i = \{+1, -1, +1, -1, +1, -1\}$ , in the stripy phase  $s_i = \{\pm 1, \mp 1, \mp 1, \pm 1, \mp 1, \mp 1\}$ , where half of the hexagons assume the upper sign, and half the lower, and in the zigzag phase  $s_i = \{+1, +1, +1, -1, -1, -1\}$ . Diagonalizing the Hamiltonian, we find the energy of the upper orbital to be, respectively,  $\sqrt{4t^2 + \Delta_{ex}^2}$ ,  $t/2 + \sqrt{9t^2 + 4t\Delta_{ex} + 4\Delta_{ex}^2}/4 + \sqrt{9t^2 - 4t\Delta_{ex} + 4\Delta_{ex}^2}/4$  (after the averaging over both types of hexagons) and  $\sqrt{5t^2/2 + \Delta_{ex}^2} + t\sqrt{9t^2/4 + 8\Delta_{ex}^2}$ . The ferromagnetic state in this model has no energy gain if  $\Delta_{ex} \leq t/2$ , and  $t - 2\Delta_{ex}$  otherwise. An elementary calculation shows that the zigzag configuration is always the lowest among antiferromagnetic states (Fig. 7). Ferromagnetism is competitive with the zigzag pattern, the actual winner being defined by a fine balance between the Hund's  $J$  and the one-electron energy scales  $t$  and SO. Indeed in the DFT calculations the energies of these two states are very close. Such frustration, combined with high itinerancy inside the hexagons, suggests that the magnetic long range order ( $T_N$ ) and the spin response (Curie-Weiss temperature  $\Theta_{CW}$  in susceptibility) may be decoupled in this system, and indeed the measured  $\Theta_{CW}$  is much larger than  $T_N$ .

This picture is, strictly speaking, applicable for the fully isolated QMO model, which, as we know, is a good approximation for the three lower bands,  $A_{1g}$  and  $E_{2u}$ , but not as good for the upper three,  $E_{2g}$  and  $B_{1u}$ . Indeed in the DFT calculations without spin-orbit the order of the magnetic states differs: the ground state is ferromagnetic, the Néel state is (as in the model) the least stable, but the zigzag and the stripy phase come out essentially



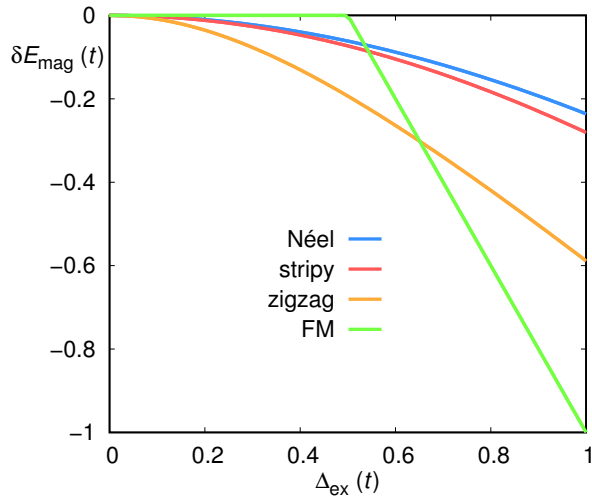


FIG. 7: One-electron DFT energy gain in the QMO model with respect to the non-magnetic energy, as a function of the exchange splitting, for various magnetic configurations.

degenerate. It is easy to understand why the ferromagnetic state is overstabilized in the DFT calculations: the Stoner energy in DFT includes the unphysical Hund's rule coupling of each orbital to itself.

Introducing the spin-orbit coupling in the DFT calculations, interestingly, makes the system somewhat closer to the above model. Even though the upper bands are now hybridized, they are well separated and thus are better described by the Hamiltonian of Eq. (3). Additionally, the zigzag phase, as opposed to the stripy one, retains the gap at the Fermi level, which gives it an additional energy advantage. As a result, the calculations with spin-orbit coupling firmly place the zigzag structure below the stripy one, although the ferromagnetic state is still competitive with the zigzag structure, for the reason discussed above [21]. Adding an atomic  $U$  makes the magnetic moments and gaps larger but does not change the energy hierarchy (we remind the reader that a physically meaningful accounting for a Hubbard  $U$  should be done for the QMOs, not the atomic orbitals).

To summarize, we have constructed a complete model of the electronic structure of a system of  $t_{2g}$  orbitals on a honeycomb lattice (of which  $\text{Na}_2\text{IrO}_3$  is an example) in the case where the dominant metal-metal hopping is the n.n. hopping *via* oxygen and this hopping is the resulting band width is the largest energy scale in the system (which is the case in  $\text{Na}_2\text{IrO}_3$ ). The model appears to be highly nontrivial and very remote from the popular Heisenberg-Kitaev model. The main factor defining the electronic structure is formation of quasi-molecular composite orbitals[22], akin to the textbook molecular orbitals of benzene or a  $\text{SH}_6$  ring. In the first approximation, the QMOs on different hexagons are orthogonal, therefore the material forms already in non-relativistic

calculations a system of narrow ( $\sim 200$  meV) subbands. The spin-orbit interaction is basically quenched for the lower two subbands, while for the upper two it leads to an additional interaction between the quasi-molecular composite orbitals, and emphasizes further the splitting of the  $t_{2g}$  band into four subbands. In the paramagnetic state, as well as in the antiferromagnetic zigzag state, only the uppermost band is empty, thus forming a band insulator. Correlation effects, while weak in terms of the ratio of the atomic  $U$  over the total  $t_{2g}$  band width, are moderately strong in terms of QMOs, thanks to the small band width of the molecular subbands. An analysis of the one-electron energy of the QMOs in the presence of an antiferromagnetic order reveals that, in agreement with the experiments, the so-called zigzag order is energetically favorable over the two competing antiferromagnetic orders, Néel and stripy, by a substantial margin. Our results open a completely new window on this intriguing compound and hopefully will lead to new theoretical and experimental research, now on the solid basis of a well-understood electronic structure. In particular, it will be instructive to address the superexchange between the QMO in the same way it is usually done for atomic orbitals; this is a subject for future work.

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# Formation of molecular orbitals and suppression of spin-orbit coupling in $\text{Na}_2\text{IrO}_3$

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The electronic structure calculations for  $\text{Na}_2\text{IrO}_3$  shown in this supplement were performed with the full potential local orbital (FPLO) code<sup>1</sup> using the generalized gradient approximation functional in its PBE form<sup>2</sup> and based on the  $C2/m$  structure as given in Ref. 3.

Fig. 1 shows the band structure of ferromagnetic  $\text{Na}_2\text{IrO}_3$  without taking spin-orbit interaction into account. The high symmetry points used in Figure 1 and in all further band structures are  $R = (1/2, 1/2, 1/2)$ ,  $X = (1/2, 0, 0)$  and  $M = (1/2, 1/2, 0)$ , given in units of the reciprocal lattice vectors.

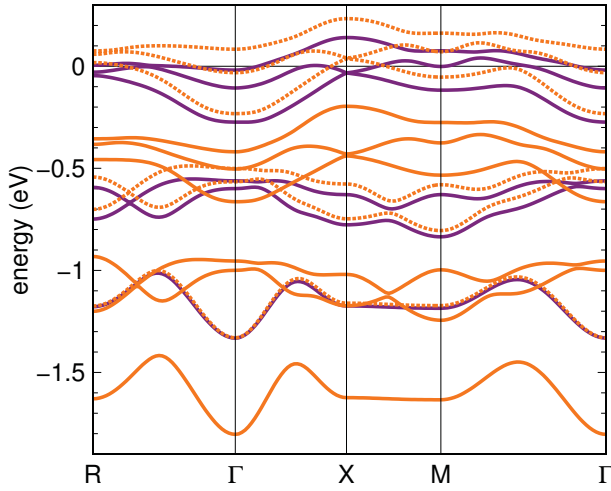


FIG. 1: Band structure of the ferromagnetic  $\text{Na}_2\text{IrO}_3$  without including spin-orbit interaction. The two line types indicate the two polarizations. Purple lines show the parent nonmagnetic band structure. The Fermi levels in both cases were aligned at  $E = 0$ .

We use projective Wannier functions<sup>4</sup> to determine a tight binding (TB) representation for the Ir  $5d$  bands. Figure 2 shows the DFT band structure together with the bands corresponding to the Wannier representation and the TB bands derived from this representation.

Figure 3 shows projective Wannier functions for the  $5d$  orbitals of one Ir site. The Wannier functions exhibit the typical shape of the  $5d$  functions at the Ir site. Besides, they show a clear asymmetry due to Na as well as tails on the O sites.

Figure 4 shows the band structure that results if we restrict the tight binding Hamiltonian to first neighbours

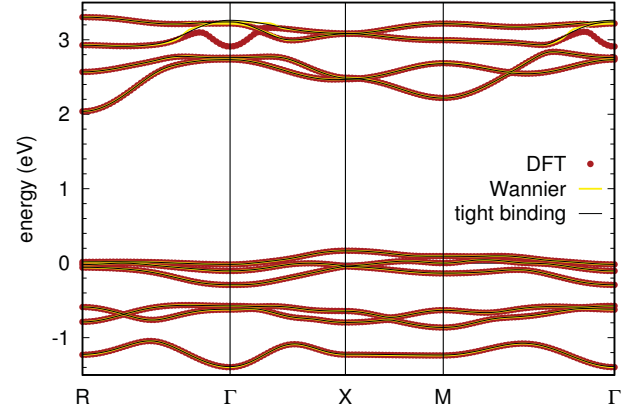


FIG. 2: Band structure of  $\text{Na}_2\text{IrO}_3$  (red symbols) shown together with the Wannier bands (yellow) and the tight binding bands (blue).

(top left), up to second nearest neighbours (top right), up to third nearest neighbours (bottom left), and without restriction (bottom right).

In Figure 5 we show the tight binding band structures that are restricted to tight binding parameters which contribute to the quasi-molecular orbitals. The four panels represent onsite terms only (onsite energy and trigonal distortion, top left), up to first nearest neighbours (top right), up to second nearest neighbours (bottom left) and up to third nearest neighbours (bottom right). Note that the small dispersion that arises for nearest neighbours is due to deviations from the perfect octahedral environment of iridium. This is most likely also the reason why for second nearest neighbours the ideal 1-2-2-1 degeneracy is lost near the Fermi level.

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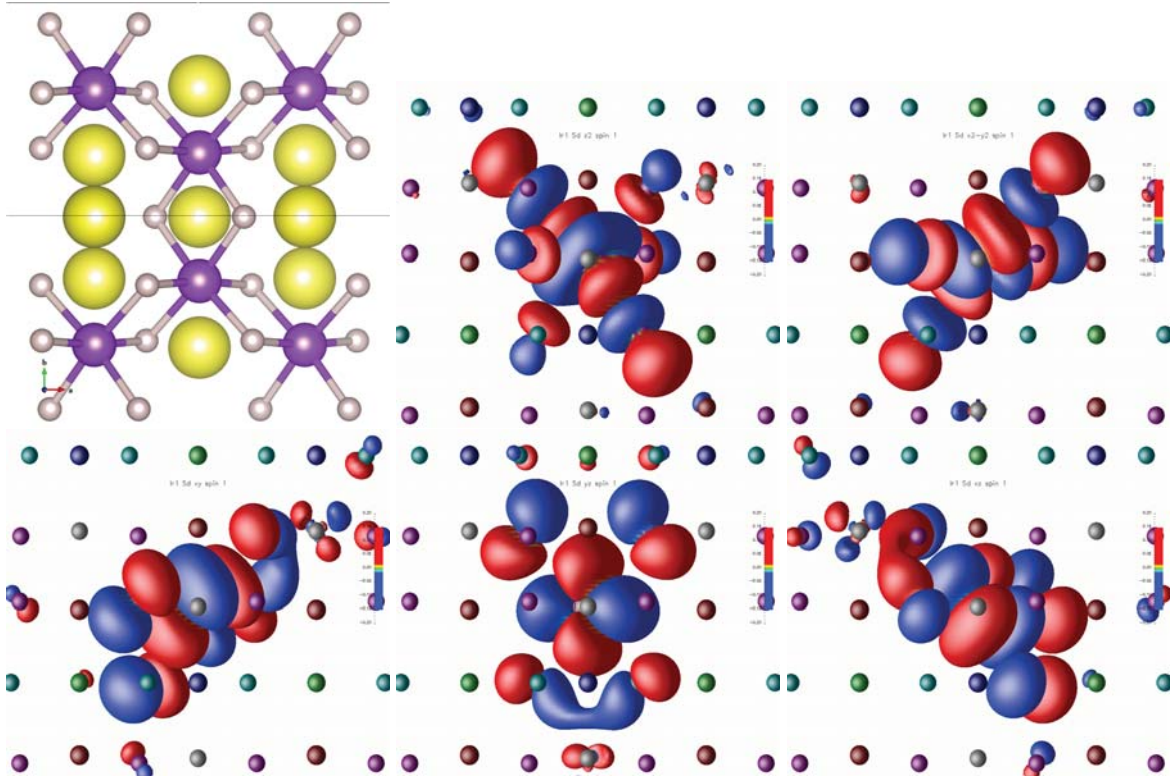


FIG. 3: Projective Wannier functions for five of the ten Ir 5d bands, together with a structure showing the perspective.

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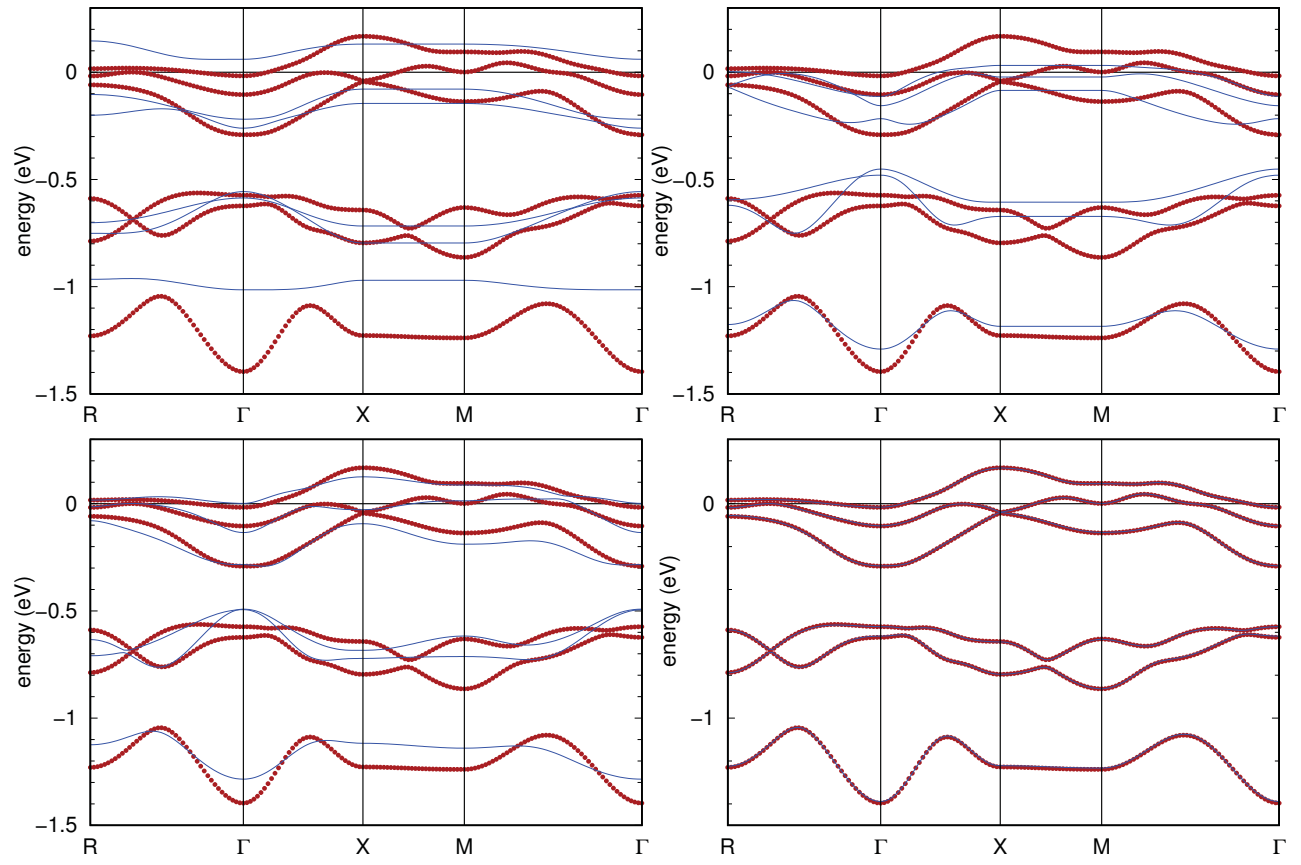


FIG. 4: Band structure of  $\text{Na}_2\text{IrO}_3$  (red symbols) shown together with the tight binding models that include only nearest neighbours (top left), up to next nearest neighbours (top right), up to third nearest neighbours (bottom left) and neighbours up to 16 Å (bottom right).

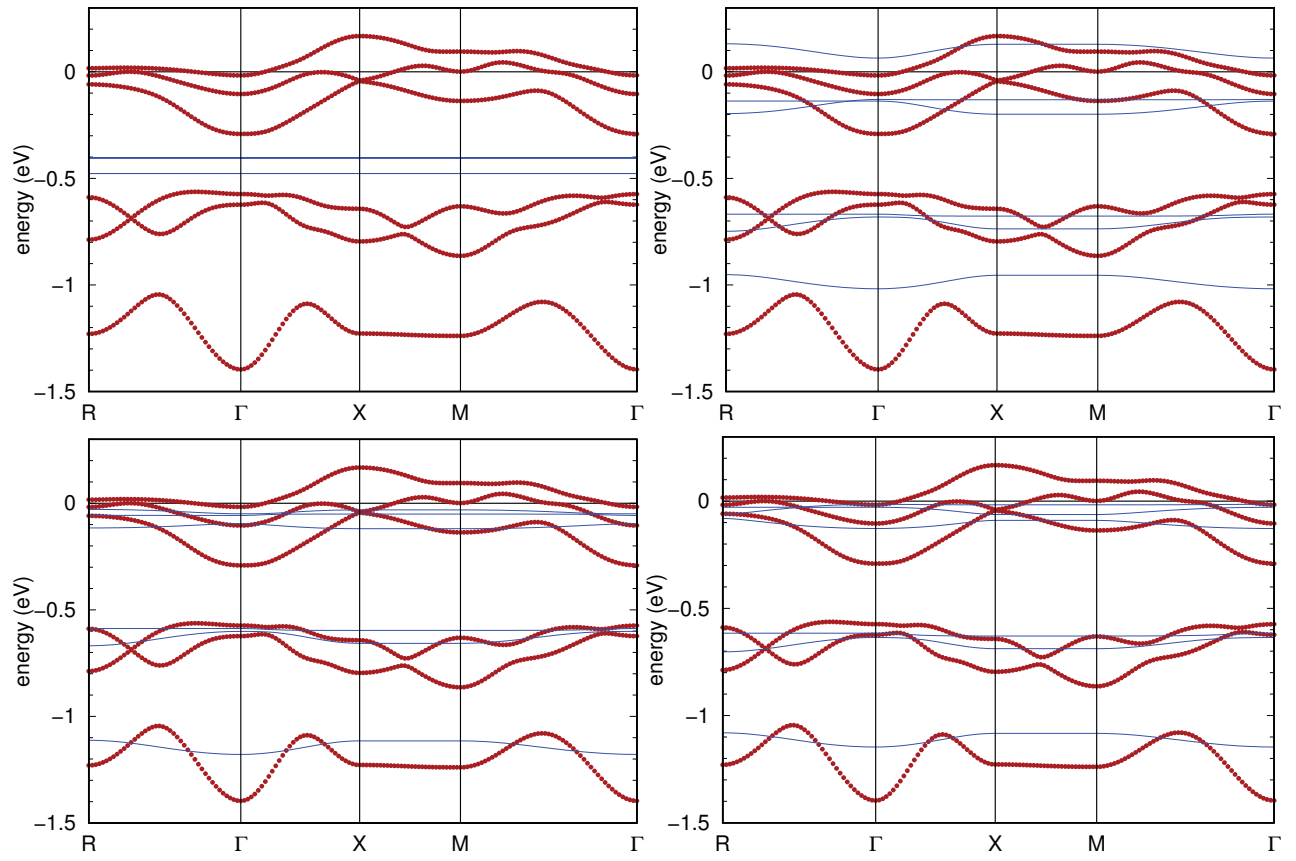


FIG. 5: Band structure of  $\text{Na}_2\text{IrO}_3$  (red symbols) shown together with the tight binding models that involve only parameters compatible with the quasi-molecular orbitals. Only on-site parameters (top left), up to nearest neighbours (top right), up to second nearest neighbours (bottom left) and up to third nearest neighbours (bottom right).